

Sheng-Li Hu,<sup>a,b</sup> Guo-Dong Yin<sup>a</sup>  
and An-Xin Wu<sup>a\*</sup><sup>a</sup>Key Laboratory of Pesticide and Chemical  
Biology of Ministry of Education, College of  
Chemistry, Central China Normal University,  
Wuhan 430079, People's Republic of China,  
and <sup>b</sup>Department of Chemistry & Environmental  
Engineering, Hubei Normal University,  
Huangshi 435002, People's Republic of ChinaCorrespondence e-mail:  
chwuax@mail.ccn.u.edu.cn

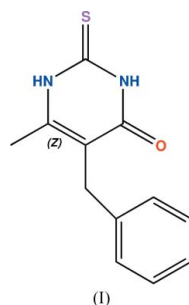
## Key indicators

Single-crystal X-ray study  
 $T = 292$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å  
 $R$  factor = 0.048  
 $wR$  factor = 0.123  
Data-to-parameter ratio = 15.6For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.5-Benzyl-6-methyl-2-thioxo-2,3-dihydro-  
pyrimidin-4(1H)-one

In the title compound,  $\text{C}_{12}\text{H}_{12}\text{N}_2\text{OS}$ , the dihedral angle between the benzene and pyrimidine rings is  $84.1(1)^\circ$ . The molecules are linked by centrosymmetric pairs of  $\text{N}-\text{H}\cdots\text{O}$  [ $\text{N}\cdots\text{O} = 2.817(2)$  Å] and  $\text{N}-\text{H}\cdots\text{S}$  [ $\text{N}\cdots\text{S} = 3.357(2)$  Å] hydrogen bonds to form chains.

## Comment

Pyrimidine derivatives have received much attention due to their versatile uses. They have been widely used as tools to study biochemical systems and as chemotherapeutic agents (Koppel *et al.*, 1961; Furberg & Petersen, 1972). As part of a general programme towards the synthesis of powerful chemotherapeutic agents, the title compound, (I), a new pyrimidine derivative, was synthesized. We report here the crystal structure of (I) (Fig. 1).



In compound (I), the benzene ring forms a dihedral angle of  $84.1(1)^\circ$  with the pyrimidine ring. The torsion angles  $\text{C}5-\text{C}6-\text{C}7-\text{C}8$  and  $\text{C}6-\text{C}7-\text{O}8-\text{C}9$  are  $160.0(2)$  and  $96.4(3)^\circ$ , respectively. The  $\text{C}=\text{O}$  and  $\text{C}=\text{S}$  bond lengths are  $1.234(3)$  and  $1.668(2)$  Å, respectively. Selected bond lengths and angles are listed in Table 1. The molecules are linked by  $\text{N}-\text{H}\cdots\text{O}$  and  $\text{N}-\text{H}\cdots\text{S}$  hydrogen bonds (Table 2) to form chains (Fig. 2).

## Experimental

The title compound was prepared according to the method described by Craig *et al.* (2000). Single crystals suitable for X-ray diffraction were obtained by slow evaporation of an acetone solution at 283 K

## Crystal data

 $\text{C}_{12}\text{H}_{12}\text{N}_2\text{OS}$   
 $M_r = 232.30$   
Monoclinic,  $P2_1/n$   
 $a = 5.8779(9)$  Å  
 $b = 29.800(4)$  Å  
 $c = 7.2160(11)$  Å  
 $\beta = 113.622(2)^\circ$   
 $V = 1158.1(3)$  Å<sup>3</sup>  
 $Z = 4$  $D_x = 1.332$  Mg m<sup>-3</sup>  
Mo  $K\alpha$  radiation  
Cell parameters from 1539  
reflections  
 $\theta = 2.7-23.4^\circ$   
 $\mu = 0.26$  mm<sup>-1</sup>  
 $T = 292(2)$  K  
Block, colourless  
 $0.20 \times 0.08 \times 0.06$  mm

## Data collection

Bruker SMART APEX CCD area-detector diffractometer  
 $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Sheldrick, 1997)  
 $T_{\min} = 0.950$ ,  $T_{\max} = 0.985$   
 5152 measured reflections

2274 independent reflections  
 1753 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.026$   
 $\theta_{\text{max}} = 26.1^\circ$   
 $h = -7 \rightarrow 4$   
 $k = -32 \rightarrow 36$   
 $l = -8 \rightarrow 8$

## Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.048$   
 $wR(F^2) = 0.123$   
 $S = 1.03$   
 2274 reflections  
 146 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.058P)^2 + 0.2016P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.20 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.16 \text{ e } \text{\AA}^{-3}$

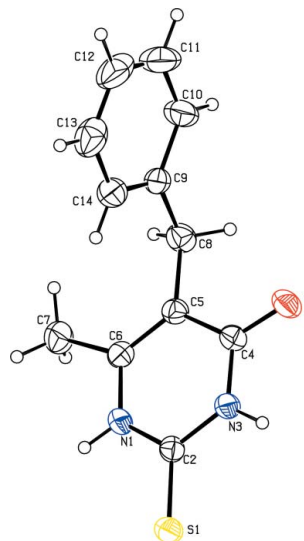


Figure 1

A view of the molecule of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level.

Table 1

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

C2–N1	1.346 (3)	C5–C8	1.497 (3)
C2–S1	1.668 (2)	C6–C7	1.496 (3)
C4–O1	1.234 (3)	C8–C9	1.508 (3)
C5–C6	1.350 (3)		
N1–C2–N3	114.66 (19)	C5–C6–N1	120.04 (19)
N1–C2–S1	123.03 (16)	C5–C8–C9	116.11 (19)
O1–C4–N3	119.7 (2)	C14–C9–C10	118.2 (2)
O1–C4–C5–C8	−0.2 (3)	C5–C8–C9–C10	160.0 (2)
C6–C5–C8–C9	96.4 (3)	S1–C2–N1–C6	178.81 (16)

Table 2

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N3–H3 $\cdots$ O1 <sup>i</sup>	0.86	1.96	2.817 (2)	176
N1–H1 $\cdots$ S1 <sup>ii</sup>	0.86	2.51	3.3574 (19)	171

Symmetry codes: (i)  $-x, -y, -z + 1$ ; (ii)  $-x + 2, -y, -z + 2$ .

The H atoms were placed in idealized positions and constrained to ride on their parent atoms, with N–H distances of 0.86  $\text{\AA}$  and C–H distances in the range 0.93–0.97  $\text{\AA}$  [C–H = 0.93  $\text{\AA}$  for phenyl, C–H = 0.96  $\text{\AA}$  and  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$  for methyl, and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  for other H atoms]. The methyl groups were allowed to rotate freely about the C–C bond.

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 2001); software used to prepare material for publication: SHELXTL.

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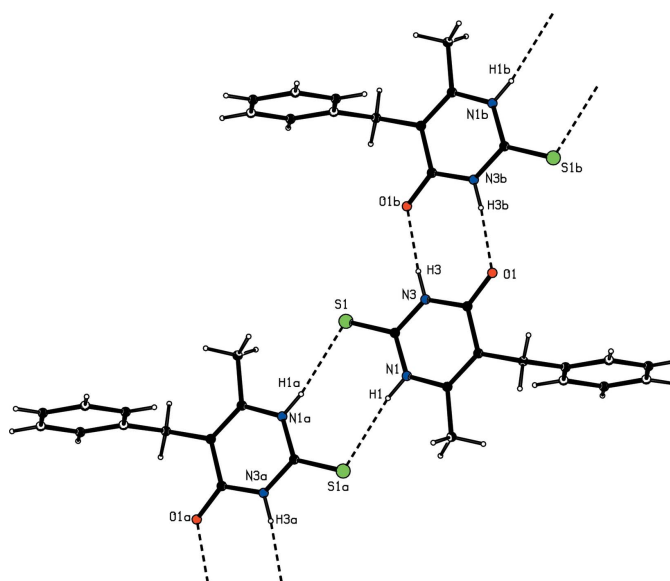


Figure 2

Part of the N–H $\cdots$ S and N–H $\cdots$ O hydrogen-bonded (dashed lines) chain in (I). [Symmetry codes: (a)  $-x, -y, -z + 1$ ; (b)  $-x + 2, -y, -z + 2$ .]

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